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Synthesis of 2-Fluoro-6-*O*-propargyl-11,12-carbamate Ketolides. A Novel Class of Antibiotics

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ABSTRACT

A novel class of 2-fluoro-6-*O*-propargyl-11,12-carbamate ketolide derivatives of erythromycin has been synthesized for antibacterial SAR studies. Replacement of the C2-hydrogen by a fluorine atom allows the synthesis of 6-*O*-propargylic ketones and electron-deficient 6-*O*-propargylic aromatic derivatives by preventing intramolecular C2-enolate Michael cyclization.

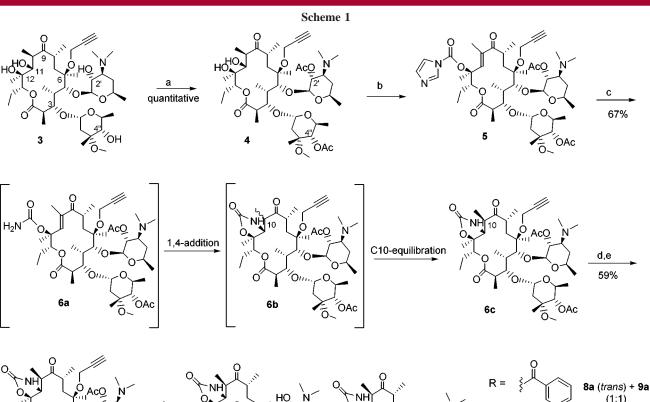
Macrolide antibiotics are a safe and effective drug class for the treatment of respiratory tract infections. Erythromycin A (1, Figure 1), a 14-membered ring macrolide, has been widely prescribed for more than 40 years. A major drawback to erythromycin is its instability in the acidic medium of the stomach, resulting in GI irritation² and poor bioavailability. To minimize the acid instability, which is associated with intramolecular ketalization of the hydroxyl groups at C6 and C12 with the ketone at C9, selective methylation of the 6-hydroxyl has been performed. The resulting 6-*O*-methyl compound, clarithromycin (2), has become a leading second generation macrolide antibiotic. However, due to the recent emergence of antibiotic resistance, macrolides with activity against resistant bacteria are needed.

Figure 1.

Subsequent modifications of clarithromycin have led to compounds with enhanced activity against resistant pathogens. In particular, the incorporation of 11,12-carbamate and 3-keto (known as ketolide) structural features has shown promise.⁵ Recent reports have further demonstrated that potent antibacterial activity against resistant strains can be achieved by introducing aromatic groups into the macrolide.⁶

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(a) Ac₂O (2.4 equiv), Et₃N, DMAP, CH₂Cl₂, rt, 24h. (b) NaH, CDI, THF/DMF (1:1), rt, 24h. (c) NH₃(aq), CH₃CN, rt, 24h.

(d) 2N HCI/EtOH, rt, 24h. (e) (i) NCS, Me₂S, CH₂CI₂,-10°C, 2h; (ii) Et₃N.

(f) 8-chloropyridone or benzoyl chloride, (Ph₃P)₂PdČl₂, Cul, Et₃N, CH₃CN, 80°C, 24h. (g) MeOH, rt, 24h.

We have been investigating novel erythromycin A derivatives that contain these key structural features as a strategy for overcoming macrolide resistance. Selective alkylation of the 6-hydroxyl in **1** with an allyl or propargyl group has recently been achieved in our laboratories, ⁷ thus providing versatile intermediates for further derivatizations.

In this Letter, we report the first synthesis of 2-F-6-*O*-propargyl-11,12-carbamate ketolide derivatives which are potent antibacterial agents. Our initial approach was to utilize the 6-O-propargyl side chain in **7** (Scheme 1) as a point of

attachment for a variety of aryl functionalities. Compound 7 was readily prepared in five steps from 6-O-propargyl erythromycin A (3).7 The synthesis began with selective protection of the 2',4"-hydroxyls of 3 as the bis-acetate 4,8 which was converted to 12-O-acyl imidazolide 5 by reacting 4 with NaH and excess 1,1'-carbonyl diimidazole. Treatment of crude 5 with liquid ammonia in acetonitrile gave carbamate 6a. Upon prolonged exposure to ammonia, 6a underwent an intramolecular Michael addition to give 6b which subsequently equilibrated to 6c exclusively. Hydrolysis of the cladinose sugar with 2 N HCl in ethanol followed by Corey—Kim oxidation¹⁰ of the resulting 3-hydroxy analogue provided ketolide 7. Attempted arylation of the propargyl side chain of 7 with 8-chloropyridone¹¹ or acylation with benzoyl chloride using Sonogashira conditions¹² produced unexpected results. Cyclization products 8a and 8b were formed as a consequence of an intramolecular C2 enolate

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^{(8) 2&#}x27;-O-Benzoate was also prepared and carried through the synthetic sequence with similar result.

Michael addition. A 1:2 mixture of *cis/trans* isomers was obtained for **8b**. For **8a**, only *trans*- α , β -unsaturated ketone was isolated along with its β , γ -unsaturated ketone **9a** in a 1:1 mixture. A propensity for these undesired reactions limited our ability to explore electron-deficient aromatic systems.

To avoid intramolecular cyclization, a C2 blocking strategy was pursued. By replacing the C2 hydrogen with a fluorine atom, undesirable enolization of the β -ketoester was prevented. Conversion of propargyl ketolide **7** to the corresponding 2-F derivative **10** by sequential treatment with NaH followed by *N*-fluorobenzene sulfonimide¹³ provided **10** as a single diastereomer (Scheme 2).¹⁴ The stereochemistry of

(a) (i) NaH (2 equiv), DMF, 0°C, 1 h., (ii) (PhSO₂)₂NF (1.1 equiv), -5°C, 3 h.

the fluorine was determined from an X-ray crystal structure of coupling product 11f (Figure 2). Furthermore, the X-ray

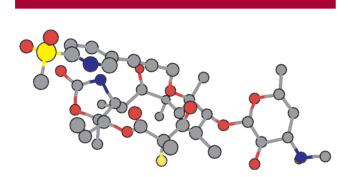


Figure 2. X-ray crystal structure of 11f.

structure of **11f** revealed the conformation of the 6-O-propargylic side chain to be in the proximity of the C2-carbon of the β -ketoester.

Sonogashira coupling of various aryl¹⁵ or acyl halides with **10** followed by deacylation of the 2'-protecting group provided modest to good yields of the desired products (Table

Table 1. Sonogashira Coupling of 2-F-6-O-propargyl Ketolide

(a) R-X, (Ph₃P)₂PdCl₂, CuI, Et₃N, CH₃CN, 80°C.

(b) MeOH, rt

entry	X	R	yield (%)
1	Cl	F 0 0 0 11a	14
2	Cl	11b	37
3	Cl	s 11c	45
4 ¹⁵	Br	N 11d	74
5 ¹⁵	Br	11e	80
6 ¹⁷	Br	N SO ₂ Me	50

1). The deacylation was readily accomplished by stirring in methanol at room temperature, presumably due to neighboring group participation of the desosamine nitrogen.¹⁶ Pre-

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liminary evaluations have revealed promising biological profiles for these compounds. Thus, the scope of our ability to explore diverse acetylenic aromatic side chains at the C6-position has been expanded.

In summary, an efficient synthesis of 2-F-6-O-propargyl-

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- (15) The C2-H analogues of **11d** and **11e** have also been prepared by coupling with **7** without intramolecular cyclization problems. Only products that were prone to Michael addition were susceptible toward intramolecular cyclization.
- (16) With the 2'-O-benzoate protecting group, the deprotection occurred in MeOH at room temperature with a longer reaction time, 1-3 days, or at reflux in 3-4 h.
- (17) The starting material has a 2'-O-benzoate protecting group in place of the 2'-O-acetate.

11,12-carbamate ketolide (10) was developed which provided a versatile intermediate for the incorporation of a variety of aryl or acyl groups which were otherwise inaccessible. The biological data for this important new class of macrolides will be reported in due course.

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Supporting Information Available: X-ray crystallographic analysis of compound **11f**. This material is available free of charge via the Internet at http://pubs.acs.org.

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